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Province of **British Columbia**

Ministry of Energy, Mines and Petroleum Resources

Geological Survey Branch Mineral Resources Division Applied Geochemistry Subsection

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NATIONAL GEOCHEMICAL RECONNAISSANCE 1:250 000 MAP SERIES

SUMDUM - TELEGRAPH CREEK, BRITISH COLUMBIA (NTS 104F - 104G)

GSC Open File 1646 **MEMPR BC RGS 19**

1987 Regional Geochemical Survey 104F - Sumdum and 104G - Telegraph Creek

BC MEMPR RGS 19, GSC OF 1646 ERRATA

Page 2, Paragraph 1, Line 2

.... samples were collected from 1220 sites.

Page 3, Table title

... (N=1286)

Page 8, Paragraph 1

2) REPEAT ANALYSES ON A SECOND SAMPLE ARE PERFORMED ON 18% (N=236) OF THE SAMPLES. SIXTY-THREE ARE RANDOMLY SELECTED AND THE REMAINDER (N=173) ARE SAMPLES HAVING VALUES THAT ARE STATISTICALLY ABOVE APPROXIMATELY THE 85TH PERCENTILE (>20 Ppb Au) OF THE TOTAL DATA SET.

Page 17 to 68

Sample types given as 9 should be 6.

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NATIONAL GEOCHEMICAL RECONNAISSANCE 1:250,000 MAP SERIES

SUMDUM - TELEGRAPH CREEK, BRITISH COLUMBIA (NTS 104F - NTS 104G)

MEMPR BC RGS 19, GSC OF 1646

1988

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1987 Regional Geochemical Survey 104F - Sumdum and 104G - Telegraph Creek

British Columbia Regional Geochemical Survey RGS-19 Geological Survey of Canada Open File 1646 National Geochemical Reconnaissance 1:250 000 Map series NGR-111

INTRODUCTION

Regional stream sediment and water geochemical reconnaissance data of northwestern British Columbia is one of three open files released in 1988 (RGS 18,19 and 20) covering NTS 104B+C, 104F+G and 104K respectively.

The 1987 reconnaissance survey was undertaken by the British Columbia Ministry of Energy, Mines and Petroleum Resources in conjunction with the Geological Survey of Canada. Funding was provided in part by the GSC under a "Letter of Understanding" for a co-operative project and by the Canada -British Columbia Mineral Development Agreement (1985 - 1989).

- P.F. Matysek, directed the British Columbia Ministry of Energy, Mines and Petroleum Resources activities.
- E.H.W. Hornbrook directed Geological Survey of Canada activities.
- P.W.B. Friske coordinated the activities of the Geological Survey of Canada staff.

Contracts let for collection, sample preparation and analysis were the responsibility of the staff of the Applied Geochemistry Subsection of the British Columbia Ministry of Energy, Mines and Petroleum Resources (MEMPR) or the Exploration Geochemistry Subdivision (GSC) as follows:

COLLECTION - McElhanney Engineering Services Limited, Vancouver, B.C.

- J.L. Gravel (MEMPR)

SAMPLE - Golder Associates, Ottawa, Ont.

PREPARATION - J.J. Lynch (GSC)

ANALYSIS - Bondar Clegg and Company Ltd., Ottawa (stream sediments)

- Chemex Labs Ltd., North Vancouver, B.C. (stream waters & gold)

- J.J Lynch (GSC) and P.F. Matysek (MEMPR)

DATA - Elan Data Makers Ltd., Victoria, B.C. (keypunching data cards)

PREPARATION - British Columbia Geological Survey (data compilation and verification)

- S.J. Day, P.F. Matysek and J.L. Gravel (MEMPR)

OPEN FILE - P.F. Matysek, S.J. Day, J.L. Gravel and W. Jackaman (MEMPR)
PRODUCTION

EGIONAL STREAM SEDIMENT AND WATER DATA, BRITISH COLUMBIA, 1987, BC RGS 19, GSC OF 1646, NTS 104F, 104G - SUMDUM, TELEGRAPH CREEK

SAMPLE COLLECTION

Helicopter (97%) and truck (3%) supported sample collection was carried out during the summer of 1987. Stream sediment and water samples were collected from 1281 sites at an average density of one site per 13.8 square kilometres throughout the 16,850 square kilometres of NTS map sheets 104F and 104G in northwestern British Columbia. Stream sediment samples ideally comprise 2 - 4 kg of sand size and finer inorganic material collected from low energy sites within the stream. Due to the rapid paced style of RGS sampling, large (6" X 5" with 2" gussets) Kraft paper sample bags were used to ensure sufficient (>40 gm) minus 80 mesh (<177 microns) fines were collected. Stream waters were taken using 250 ml nalgene plastic bottles flushed out with water from the stream prior to collection of the sample. Duplicate samples were routinely collected from a site once in each analytical block of twenty samples. To aid in the follow-up of survey results, highly visible aluminum tags (5 by 10 centimetres) bearing a unique RGS sample number were used to mark every sample site.

SAMPLE PREPARATION

Sediment samples were air dried, first on open air racks, and then within a heated (50C) drying shed. Dried samples were sieved to minus 18 mesh (approximately 1 mm) to reduce sample weight and to determine the fines content. Sample quality checks were ran by routinely sieving to minus 80 mesh (<177 microns) 1 sample in each block of 20, plus any samples suspected of low fines content. Samples found to be deficient in fines (<40 gm), were resampled. Field prepared samples were then shipped to Golder Associates for further sieving to minus 80 mesh (<177 microns) and then ball milling. At this time, control reference samples and blind duplicate samples were inserted into each block of twenty sediment samples. For the water samples, only control reference samples were inserted into the block. There were no blind duplicates for water samples.

ANALYTICAL DETERMINATIONS

Stream Sediments

The prepared minus 80 mesh (<177 microns) fraction of sediment samples were sent to Bondar Clegg and Company Ltd. of Ottawa for analytical determinations of elements listed in Table 1. Chemex Labs of North Vancouver performed the analyses for gold.

Stream Waters

Water samples were sent to Chemex Labs of North Vancouver, B.C. for analysis of elements listed in Table 2.

GIONAL STREAM SEDIMENT AND WATER DATA, BRITISH COLUMBIA, 1987, BC RGS 19, GSC OF 1646, NTS 104F, 104G - SUMDUM, TELEGRAPH CRE

TABLE 1 - Stream Sediment Analyses (N=1291)

Element	Units	Method Detection Limit	No. of Samples Detection at Limit	Percentage at Detection Limit
\Zinc	PPM	ATOMIC ABSORPTION SPECTROSCOPY (AAS) < 2 = 1	0	0.0
Copper	PPM	ATOMIC ABSORPTION SPECTROSCOPY $< 2 = 1$	12	0.9
v./Lead	PPM	ATOMIC ABSORPTION SPECTROSCOPY $< 2 = 1$	19	1.5
Nickel	PPM	ATOMIC ABSORPTION SPECTROSCOPY $< 2 = 1$	48	3.7
⊮ Cobalt	PPM	ATOMIC ABSORPTION SPECTROSCOPY $< 2 = 1$	39	3.0
Silver	PPM	ATOMIC ABSORPTION SPECTROSCOPY $< 0.1 = 0.1$	1 1113	86.2
Managanese	PPM	ATOMIC ABSORPTION SPECTROSCOPY $< 5 = 5$	0	0.0
Arsenic	PPM	HYDRIDE EVOLUTION AAS $< 1 = 1$	258	20.0
Molybdenum	PPM	ATOMIC ABSORPTION SPECTROSCOPY $< 1 = 1$	1056	81.8
WIron	PCT	ATOMIC ABSORPTION SPECTROSCOPY $< 0.02 = 0$.	01. 0	0.0
Mercury	PPB	FLAMELESS AAS $< 5 = 5$	359	27.8
LOI	PCT	LOSS ON IGNITION (WEIGHT DIFFERENCE) $< 0.2 = 0$.	1 198	15.3
V/Uranium	PPM	DELAYED NEUTRON ACTIVATION $< 0.5 = 0$.	1 6	0.5
Fluorine	PPM	SPECIFIC ION ELECTRODE $< 40 = 20$	0	0.0
Vanadium	PPM	ATOMIC ABSORPTION SPECTROSCOPY $< 5 = 5$	2	0.2
//Cadmium	PPM	ATOMIC ABSORPTION SPECTROSCOPY $< 0.2 = 0.$	1 917	71.0
Antimony	PPM	HYDRIDE EVOLUTION AAS $< 0.2 = 0.$	1 414	32.1
Tungsten	PPM	COLORIMETRY $< 2 = 1$	1206	93.4
Barium	PPM	ENERGY DISPERSIVE XRF < 20 = 10	0	0.0
//Tin	PPM	ATOMIC ABSORPTION SPECTROSCOPY $< 2 = 1$	343	26.6
/ Gold	PPB	FIRE ASSAY - NAA Variable = 1	574	44.5

TABLE 2 - Water Analyses

Element	Units	Method	Detection Limit
/Uranium Fluorine pH	PPB PPB LOG	LASER-INDUCED FLUOROMETRY SPECIFIC ION ELECTRODE SPECIFIC ION ELECTRODE	< 0.05 = 0.02 < 20 = 10

Tables 1 and 2 display the detection limits of the analytical methods used to determine the above elements. The figure to the right of the detection limit heading corresponds to an arbitarilily set value if the results fall below the contracted commercial laboratory's detection limit (usually 1/2 the detection limit) and are used for the mathematical calculations and the listings.

EGIONAL STREAM SEDIMENT AND WATER DATA, BRITISH COLUMBIA, 1987, BC RGS 19, GSC OF 1646, NTS 104F, 104G - SUMDUM, TELEGRAPH CRI

ANALYTICAL METHODS Stream Sediments

For the determination of Zn, Cu, Pb, Ni, Co, Ag, Mn, Fe, Cd and As in stream sediments a 1 gram sample was reacted with 3 ml conc. HNO3 in a test tube overnight at room temperature. After digestion, the test tube was immersed in a hot water bath at room temperature and brought up to 90C and held at this temperature for 30 minutes with periodic shaking. 1 ml conc. HCL was added and heating was continued for another 90 minutes. The sample solution was diluted to 20 ml with metal-free water, mixed and allowed to stand for two hours.

Zn, Cu, Pb, Ni, Co, Ag, Mn, Fe and Cd were measured by aspirating the test solution into an atomic absorption spectrophotometer using an air-acetylene flame. Background corrections were made for Pb, Ni, Co, Ag and Cd.

As was determined using a hydride evolution method wherein the hydride (ASH₃) is evolved and passed through a heated quartz tube in the light path of an atomic absorption spectrophotometer. The method is described by Aslin (1976).

Molybdenum and Vanadium were determined by atomic absorption spectroscopy using a nitrous oxide acetylene flame. A 0.5 gram sample was reacted with 1.5 ml concentrated HN03 at 90C for 30 minutes. At this point 0.5 ml concentrated HCL was added and the digestion was continued at 90C for an additional 90 minutes. After cooling, 8 ml of 1250 ppm Al solution were added and the sample solution was diluted to 10 ml before aspiration.

Mercury was determined using a modified Hatch and Ott procedure. The method is described by Jonasson et al. (1973). A 0.5 gram sample was reacted with 20 ml concentrated HNO3 and 1 ml concentrated HCL in a test-tube for 10 minutes at room temperature prior to 2 hours of digestion with mixing at 90C in a hot water bath. After digestion, the sample solutions were cooled and diluted to 100 ml with metal free water. The Hg present was reduced to the elemental state by the addition of 10 ml 10% w/v SnSO4 in 1M H2SO4. The Hg vapour was then swept by a stream of air into an absorption cell mounted in the light path of an atomic absorption spectrophotometer. Absorption measurements were made at 253.7 nm.

Loss on Ignition was determined using a 500 mg sample. The sample, weighed into 30 ml beaker, was placed in a cold muffle furnace and brought up to 500C over a period of 2-3 hours. The sample was left at this temperature for 4 hours, then allowed to cool to room temperature for weighing.

Uranium was determined using a neutron activation method with delayed neutron counting. A detailed description of the method is provided by Boulanger et al. (1975). In brief, a 1 gram sample was weighed into a 7 dram polyethylene vial, capped and sealed. The irradiation was provided by the Atomic Energy of Canada's Slowpoke II reactor. Calibration was carried out once a day as a minimum, using natural materials of known uranium concentration.

GIONAL STREAM SEDIMENT AND WATER DATA, BRITISH COLUMBIA, 1987, BC RGS 19, GSC OF 1646, NTS 104F, 104G - SUMDUM, TELEGRAPH CRE

Fluorine was determined as described by Ficklin (1970). A 250 mg sample was sintered with 1 gram of a flux consisting of two parts by weight sodium carbonate and 1 part by weight potassium nitrate. The residue was then leached with water, the sodium carbonate was neutralized with 10 ml 10% (w/v) citric acid and the resulting solution was diluted to 100 ml with water. The pH of the resulting solution was 5.5 to 6.5. The fluoride content of the test solution was then measured using a fluoride ion electrode. Standard solutions contain sodium carbonate and citric acid in the same quantities as the sample solution. A detection limit of 40 ppm was achieved.

Antimony was determined as described by Aslin (1976). A 500 mg sample was placed in a test tube; 3 ml concentrated HNO3 and 9 ml concentrated HCL are added and the mixture was allowed to stand overnight at room temperature. The mixture was heated slowly to 90C and maintained at this temperature for at least 90 minutes. The solution was cooled and diluted to 10 ml. A 400 ul aliquot of this test solution was removed and diluted to 10 ml with 1.8 M HCL. The antimony in an aliquot of this dilute solution was then determined by hydride evolution—atomic absorption spectrometry.

Tungsten was determined as follows: A 0.2 gm sample of stream sediment was fused with 1 gm K2S2O7 in a rimless test tube at 575C for 15 minutes in a furnace. The mixture was then leached with water. Tungsten was reduced with stannous chloride and complexed with thiocyanate. The tungsten was selectively extracted into a carbon tetrachloride-tri-n-butyl phosphate mixture whereby the concentration was determined by comparison of the resulting color of the sample solution with known standards prepared at the same time.

Barium was determined in sediments using energy-dispersive X-Ray fluorescence according to the method described by Bright (1983). A 5 gm sample of sediment is placed in a 10 ml plastic vial. The sample is exposed to a radioactive source and the intensities of X-rays emitted are measured. The amount of barium is measured by comparison to standards. Corrections are made for inter-element interferences.

Tin in stream sediments was determined as follows: A 200 mg sample was heated with NH4I; the sublimed SnI4 was dissolved in acid and the tin determined by atomic absorption spectrometry.

Gold was determined usually on a 10 gram sample: Depending on the amount of sample available, lesser weights were sometimes used. This resulted in a variable detection limit: 1 ppb Au for a 10 gram sample, 2 ppb Au fo a 5 gram sample. Samples less than 5 grams were not analysed. The sample was fused to produce a lead button, collecting any gold in the sample, which was cupelled in a muffle furnace to produce a silver (dore) bead. The silver beads were

irradiated in a neutron flux for 1 hour, cooled for 4 hours, and counted by gamma ray spectrometry, calibration was carried out using standard and blank beads.

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REGIONAL STREAM SEDIMENT AND WATER DATA, BRITISH COLUMBIA, 1987, BC RGS 19, GSC OF 1646, NTS 104F, 104G - SUMDUM, TELEGRAPH CREEK

ANALYTICAL METHODS Stream Waters

Fluoride ion complexes in water samples were determined using a fluoride electrode. Prior to measurement, an aliquot of the sample was mixed with an equal volume of TISAB II solution (Total Ionic Strength Adjustment Buffer). The TISAB II buffer solution was prepared as follows: 58 gm NaCl and 5 gm CDTA (cyclohexylene dinitrilo acetic acid) were dissolved in a mixture of 50 ml metal free-water and 57 ml glacial acetic acid. The solution was cooled to room temperature and the pH adjusted to between 5.0 and 5.5 by the slow addition of 5M NaOH solution. The solution was cooled and diluted to 1 litre in a volumetric flask.

Hydrogen ion activity (pH) was measured with a combination glass-calomel electrode and a pH meter.

Uranium in waters was determined by a laser-induced fluorometric method using a scintrex UA-3 Uranium analyser. All readings were taken against a sample blank.

Quality control and monitoring of the geochemical data was undertaken by a standard method used by the Exploration Geochemistry Subdivision at the Geological Survey of Canada.

EGIONAL STREAM SEDIMENT AND WATER DATA, BRITISH COLUMBIA, 1987, BC RGS 19, GSC OF 1646, NTS 104F, 104G - SUMDUM, TELEGRAPH CRE

COMMENTS REGARDING INTERPRETATION OF GOLD RESULTS

The following discussion reviews the format used to present the gold geochemical data and outlines some important points to consider when interpreting this data. This discussion is included in recognition of the special geochemical behaviour and mode of occurrence of gold in nature and the resultant difficulties in obtaining and analyzing samples which reflect the actual concentration level at a given site.

Understanding Au geochemical data from regional stream sediment requires an appreciation of the unique chemical and physical characteristics of Au and its mobility in the surficial environment. Key properties that distinguish the geochemical behaviour of gold from most other elements include:

- AU OCCURS MOST COMMONLY IN THE NATIVE FORM. IT IS CHEMICALLY AND PHYSICALLY RESISTANT AND A HIGH PROPORTION OF THE METAL IS DISPERSED IN MICRON-SIZED PARTICULATE FORM. GOLD'S HIGH SPECIFIC GRAVITY RESULTS IN HETEROGENEOUS DISTRIBUTIONS, ESPECIALLY IN STREAM SEDIMENTS.
- 2) AU TYPICALLY OCCURS AT LOW CONCENTRATIONS IN THE PPB RANGE. AU CONCENTRATIONS OF A FEW PPM MAY REPRESENT ECONOMIC DEPOSITS. BACKGROUND LEVELS ENCOUNTERED FOR STREAM SEDIMENTS SELDOM EXCEED 10 PPB, AND COMMONLY ARE NEAR THE DETECTION LIMIT OF 1 PPB.

The many foregoing factors can result in a particle sparsity or 'nugget' effect, wherein very low concentrations of Au are heterogeneously distributed in the surficial environment. Hence, a major problem facing explorationists is obtaining a representative sample. In general, the lower the concentration of Au, the larger the sample size required to reduce uncertainty over whether subsample analytical values truly represent actual values. Conversely, as Au concentrations increase the number of Au particles to be shared in random subsamples increases and the variability of results decreases (Clifton et al., 1969; Harris, 1982).

The limited amount of material collected during the rapid, reconnaissance-style regional surveys and the need to analyze for a broad spectrum of elements, precludes the use of a significantly large sample weight for the Au analyses (usually 10.0 grams). Consequently, results from these analyses tend to be highly variable and qualitative rather than quantitative. To evaluate and monitor the sampling and analytical varibility which are inherent in the analysis of gold in geochemical mediums, the following control methods are currently employed:.

- 1) FOR EACH BLOCK OF TWENTY SAMPLES:
 - A) RANDOM INSERTION OF A STANDARD REFERENCE SAMPLE TO MONITOR AND CONTROL ANALYTICAL ACCURACY AND LONG-TERM PRECISION;
 - B) COLLECTION OF A FIELD DUPLICATE (TWO SAMPLES COLLECTED SEPARATELY FROM ONE SITE) TO MONITOR SAMPLING VARIANCE; AND
 - C) ANALYSIS OF A SECOND SUBSAMPLE (BLIND DUPLICATE) FROM ONE SAMPLE TO MONITOR AND CONTROL SHORT-TERM PRECISION.

REGIONAL STREAM SEDIMENT AND WATER DATA, BRITISH COLUMBIA, 1987, BC RGS 19, GSC OF 1646, NTS 104F, 104G - SUMDUM, TELEGRAPH CREEK

2) REPEAT ANALYSES ON A SECOND SUBSAMPLE ARE PERFORMED ON 20% (N=155) OF THE SAMPLES. THIRTY NINE SAMPLES ARE RANDOMLY SELECECTED AND THE REMAINDER ARE SAMPLES HAVING VALUES THAT ARE STATISTICALLY ABOVE APPROXIMATELY THE 85TH PERCENTILE (> 40 PPB AU) OF THE TOTAL DATA SET.

Gold data presentation, statistical treatment and the map format are somewhat different than for the other elements. Gold data listed in this open file include initial analytical results, values determined from repeat analyses, together with sample weights and corresponding detection limits for all analysed samples. The gold histogram, and statistical parameters were determined using the following data population selection criteria:

- 1) ONLY THE FIRST VALUE OF A REPEAT ANALYSIS WAS UTILIZED
- 2) AU VALUES LESS THAN THE THE VARIABLE DETECTION LIMIT WERE SET TO 1 PPB

On the gold symbol and value map, repeat analysis values (not field duplicates) for those samples initially having concentrations greater than the 85th percentile are placed in brackets following the initial value determination. Following are possible variations in the way data is presented on the gold symbol and value map:

NO POINT PLOTTED, NO GOLD DATA, INSUFFICIENT SAMPLE FOR ANALYSIS

- + 27 SINGLE ANALYSIS
- + 27 (42) REPEAT ANALYSIS

In summary, geochemical follow-up investigations should be based on a careful consideration of all geological and geochemical information, and especially a careful appraisal of gold geochemical data and its variability. In some cases, prospective follow-up areas may be indirectly identified by pathfinder element associations in favourable geology, although an anomalous Au response due to natural variability may be lacking. Once an anomalous area has been identified, field investigations should be designed to include detailed geochemical follow-up surveys and collection of large representative samples. Subsequent repeat subsample analyses will increase the reliability of results and permit a better understanding of natural variability which can then be used to improve sampling methodology and interpretation.

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EGIONAL STREAM SEDIMENT AND WATER DATA, BRITISH COLUMBIA, 1987, BC RGS 19, GSC OF 1646, NTS 104F, 104G - SUMDUM, TELEGRAPH CRI

DATA PREPARATION

Field data comprising sample identification, location, and both sample and site descriptions, were recorded during collection by the contract field crew onto field cards (REV. 74) used by the Geological Survey of Canada (Garrett, 1974). The field data were then keypunched and generated into computer files by Elan Data Makers Ltd. of Victoria, B.C. The files were subsequently verified for accuracy by the staff of the Applied Geochemistry Subsection.

Sample locations, marked on 1:50 000 scale NTS maps while in the field, were transferred to a 1:250 000 scale map in the base camp. The map was digitized by the British Columbia Geological Survey to obtain the sample site UTM coordinates needed for the production of 1:250 000 scale sample location and geochemical symbol and value maps. The five 1:100 000 scale sample location maps covering map sheet 104F+G were produced by the Applied Geochemistry Subsection using sample site coordinates digitized from the original 1:50 000 scale NTS field maps. Sample site coordinates were checked as follows: A sample location map was produced on a Houston Instruments DMP-40 plotter using the digitized coordinates; the field contractor's sample location map was then overlayed with the computer plotted map; the two sets of points were checked for coincidence. Dominant rock types in the stream catchment basins were identified using the GSC 1:1 000 000 scale Geological Atlas series map 1418A, ISKUT RIVER - Sheet 104, 114, compiled by Souther et al. (1979). A portion of this same map, enlarged and combined with the Telegraph Creek and Sumdum - 1:250 000 scale NTS maps, forms the base for the geochemical maps in this open file. Surficial geology for part of 104F and 104G derived from Ryder (1984) has been included with the geochemical maps. Computer files consisting of field observations, sample location co-ordinates, geological data and analytical data were combined into one file. Software programs developed at Applied Geochemistry Subsection by K. Talvila, S.J Day and P.F. Matysek were used to produce the detailed listings, univariate statistics, open file value and symbol maps and floppy diskettes.

EGIONAL STREAM SEDIMENT AND WATER DATA, BRITISH COLUMBIA, 1987, BC RGS 19, GSC OF 1646, NTS 104F, 104G - SUMDUM, TELEGRAPH CR

DATA LIST LEGEND - STREAMS

- MAP National Topographic System (NTS) lettered quadrangle (scale 1:50 000). part of sample number
- ID Remainder of sample number
 YEAR (first 2 digits),
 FIELD CREW (3rd digit) SAMPLE SEQUENCE NUMBER (last 3 digits)
- UTM Universal Transverse Mercator (UTM)
 COORDNATES coordinate system sample coordinates
 - ZN UTM zone
 - EAST UTM easting coordinate (in metres)
 - NORTH UTM northing coordinate (in metres)
- FORMATION Rock unit label on geology base map
- ROCK TYPE Major rock type at the sample site
 - AGE Stratigraphic age of the rock type
 - WD Width of stream at sample site (in decimetres)
 - DT Depth of stream at sample site (in decimetres)
 - SAMP Type of material sampled
 - RP ST Replicate status- relationship of sample
 to others within the block of 20
 - CONT Contamination at site or within the
 drainage basin
 - BANK Bank type
 - WCOL Water colour and suspended load

- RATE Water flow rate
- SCOL Predominant sediment colour
- SMP CMP Sample Composition- proportional volume of sand, inorganic fines and organics respectively
 - PPPS Precipitate or stain on sediments
 at sample site
 - PRPB Distinctive precipitate, stain, weathering, blooms on rocks in immediate catchment area
 - PHYS General physiography
 - PATT Drainage pattern
 - TYPE Stream type
 - CLSE Stream class
 - SRCE Source of water

GIONAL STREAM SEDIMENT AND WATER DATA, BRITISH COLUMBIA, 1987, BC RGS 19, GSC OF 1646, NTS 104F, 104G - SUMDUM, TELEGRAPH CREEK

DATA LIST LEGEND - STREAMS (CONTINUED)

SAMP:

- 1 Stream Sediment only
- 2 Spring Sediment
- 3 Heavy Mineral Concentrate
- 4 Stream Water
- 5 Spring/Well Water
- 6 Stream Sediment and Water collected at site

RP ST:

- 00 Routine Sample
- 10 First of Field Duplicate
- 20 Second of Field Duplicate

CONT:

- 0 None
- 1 Possible
- 2 Probable
- 3 Definite
- 4 Mining Activity
- 5 Industrial Sources
- 6 Agricultural
- 7 Domestic, Household Sources
- 8 Forestry
- 9 Burned areas

BANK:

- 0 Undefined
- 1 Alluvial
- 2 Colluvial
- 3 Glacial Till
- 4 Glacial Outwash
- 5 Exposed Bedrock
- 6 Talus, Scree
- 7 Organic

WCOL:

- 0 Clear
- 1 Brown Transparent
- 2 White Cloudy
- 3 Brown Cloudy

RATE:

- 0 Stagnant
- 1 Slow
- 2 Moderate
- 3 Fast
- 4 Torrent

SCOL:

- 1 Red to Brown
- 2 White to Buff
- 3 Black
- 4 Yellow
- 5 Green
- 6 Blue to Grey

SMP CMP:

- Proportional Volume of Sand, Inorganic Fines and Organics, based on thirds
- 0 Absent
- 1 Less than 1/3
- 2 Between 1/3 and 2/3
- 3 Greater than 2/3

EXAMPLES:

- 013 = 0% Sand, 25% Fines and
 - 75% Organics
- 220 = 50% Sand and 50% Fines
- 030 = 100% Fines

PPPS:

0 - None (otherwise SCOL scale)

PRPB:

0 - None (otherwise SCOL scale)

PHYS:

- 1 Muskeg, Swampland
- 2 Peneplain, Plateau
- 3 Hilly, Undulating
- 4 Mountainous, Mature
- 5 Mountainous Youthful

PATT:

- 0 Poorly Defined
- 1 Dendritic
- 2 Herringbone
- 3 Rectangular
- 5 Discontinuous

TYPE:

- 1 Permanent
- 2 Seasonal

CLSE:

- 1 Primary
- 2 Secondary
- 3 Tertiary
- 4 Quaternary

SRCE:

- 0 Unknown
- 1 Groundwater
- 2 Spring Runoff
- 3 Recent Precipitation
- 4 Glacier Melt Water

EGIONAL STREAM SEDIMENT AND WATER DATA, BRITISH COLUMBIA, 1987, BC RGS 19, GSC OF 1646, NTS 104F, 104G - SUMDUM, TELEGRAPH CR

ROCK MNEMONICS AND AGES FOR 104F - SUMDUM AND 104G - TELEGRAPH CREEK

Note 1: Rock Mnemonics must always be used in conjunction with age as the mnemonic may be repeated.

Note 2: The column with heading "FORMATION" contains the rock unit label on the open file geology base.

FORMROCK AGE ATION DESCRIPTION

STRATIFIED ROCKS

QUATERNARY, PLEISTOCENE AND RECENT

BSLT	64	Rvb	Basalts, cinder, ash
TILL	64	Qs	Surficial clastic sediments and glacial deposits
OLVB	64	Qvo	Olivine Basalt
			TERTIARY AND QUATERNARY
BSLT	63	PPLM	Level Mountain Group: basalt
BTRT	63	PPvb	Basalt, rhyolite, olvine basalt
RYLT	63	PPvr	Rhyolite, trachyte, tuff
RYLT	63	ESL	TERTIARY Sloke Croup, wheelite trackute andesite baselt
RILT	6.5	FOT	Sloko Group: rhyolite, trachyte, andesite, basalt
			CRETACEOUS AND TERTIARY
ANDS	56	KTvd	Andesite

EGIONAL STREAM SEDIMENT AND WATER DATA, BRITISH COLUMBIA, 1987, BC RGS 19, GSC OF 1646, NTS 104F, 104G - SUMDUM, TELEGRAPH CRI

ROCK	MNEMONICS	AND	AGES	FOR	104F	_	SUMDUM	AND	104G	-	TELEGRAPH	CREEK	(continued)	

			CRETACEOUS
SNDS	55	uKTC	Tango Creek: sandstone, siltstone, coal
			JURASSIC AND CRETACEOUS
SLSN	51	JKs	Upper Hazelton Group: siltstone, greywacke, conglomerate, shale
			JURASSIC
SLSN	50	JHs	Hazelton Group: siltstone, greywacke, conglomerate, shale
CGLM	49	JŦ	Takwahoni Group: conglomerate, grit, greywacke
CGGK	49	Jcg	Conglomerate, grit, greywacke
BSLT	49	mJvb	Basalt, pillow lava, tuff, volcaniclastic rocks
SHLE	49	Jр	Shale
			TRIASSIC
PLLT	45	uTp	TRIASSIC Phyllite, argillite, siltstone, greywacke, limestone
PLLT SLSN		uTp uTs	
		-	Phyllite, argillite, siltstone, greywacke, limestone
SLSN	45	uTs	Phyllite, argillite, siltstone, greywacke, limestone Siltstone, chert, sandstone, tuff
SLSN	45 45 45	uTs uTs v	Phyllite, argillite, siltstone, greywacke, limestone Siltstone, chert, sandstone, tuff Undifferentiated andesitic volcanic and clastic sedimentary rocks
SLSN ANDV VLRK	45 45 45	uTs uTs v uTST	Phyllite, argillite, siltstone, greywacke, limestone Siltstone, chert, sandstone, tuff Undifferentiated andesitic volcanic and clastic sedimentary rocks Stuhini Group: undifferentiated volcanic and sedimentary rocks
SLSN ANDV VLRK ANDS	45 45 45 45	uTsv uTsv uTST uTvd	Phyllite, argillite, siltstone, greywacke, limestone Siltstone, chert, sandstone, tuff Undifferentiated andesitic volcanic and clastic sedimentary rocks Stuhini Group: undifferentiated volcanic and sedimentary rocks Andesite, pyroclastic rocks, greenstone

SIONAL STREAM SEDIMENT AND WATER DATA, BRITISH COLUMBIA, 1987, BC RGS 19, GSC OF 1646, NTS 104F, 104G - SUMDUM, TELEGRAPH CREI

ROCK MNEMONICS AND AGES FOR 104F - SUMDUM AND 104G - TELEGRAPH CREEK (continued)

			CARBONIFEROUS AND PERMIAN
SCST	35	CPsn	Schist, gneiss
GRNS	35	CPsv	Greenstone, limestone, shale, clastic sedimentary rocks
7 hemys	24	Vah	MISSISSIPIAN
LMTF	34	Mct	Limestone, tuff, chert
		•	PLUTONIC ROCKS
			CRETACEOUS AND TERTIARY
FLSP	56	KTfp	Felsite, feldspar porphyry
GRPR	56	KTgp	Granite porphyry, granophyre, syenite
QTMZ	56	KTqm	Quartz monzonite
			CRETACEOUS
QTMZ	52	Kdw	Quartz monzonite
			JURASSIC AND CRETACEOUS
GDRD	51	JKgd	Granodiorite
DORT	51	JKdi	Diorite
QRZD	51	JKqd	Quartz diorite

EGIONAL STREAM SEDIMENT AND WATER DATA, BRITISH COLUMBIA, 1987, BC RGS 19, GSC OF 1646, NTS 104F, 104G - SUMDUM, TELEGRAPH CRE

ROCK MNEMONICS AND AGES FOR 104F - SUMDUM AND 104G - TELEGRAPH CREEK (continued)

			TRIASSIC AND JURASSIC
GRDR	46	TJgd	Granodiorite
QRZD	46	TJdi	Quartz diorite
SYNT	46	Ŧjy	Syenite, monzonite
			TRIASSIC
DORT	42	Tb	Diorite, gabbro
DORT	42	Tdi	Diorite, monzonite
			PERMIAN
UMFC	40	PTub	Ultramafic rocks, serpentinite
			AGE UNKNOWN
GRDR	65	gđ	Granodiorite
AMPH	65	m	Amphibole, gneiss, migmatite

IONAL STREAM SEDIMENT AND WATER DATA, BRITISH COLUMBIA, 1987, BC RGS 19, GSC OF 1646, NTS 104F, 104G - SUMDUM, TELEGRAPH CREE

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