REGIONAL STREAM SEDIMENT AND WATER GEOCHEMICAL DATA, BRITISH COLUMBIA 1985, GSC-OF 1214, NGR 79-1985, NTS 93G

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* OPEN FILE 1214

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GEOLOGICAL SURVEY OF CANADA OPEN FILE 1214:BRITISH COLUMBIA MINISTRY OF ENERGY, MINES AND PETROLEUM RESOURCES OPEN FILE RGS-13
REGIONAL STREAM SEDIMENT AND WATER GEOCHEMICAL RECONNAISSANCE DATA OF CENTRAL BRITISH COLUMBIA IS ONE OF THREE OPEN FILES RELEASED IN 1986 (1214,1215 AND 1216) COVERING NTS 93G,93H AND 93J RESPECTIVELY.
REGIONAL STREAM SEDIMENT AND WATER GEOCHEMICAL RECONNAISSANCE DATA FOR 93G(E1/2) AND 93H(W1/2) PREVIOUSLY RELEASED AS OPEN FILE 1107 (COLLECTED IN 1984, RELEASED IN 1985) ARE REISSUED ALONG WITH THE COMPLETED SURVEY COVERAGE OF THESE NTS SHEETS.

THE (1984) RECONNAISSANCE SURVEY WAS UNDERTAKEN BY THE GEOLOGICAL SURVEY OF CANADA (GSC) IN CONJUNCTION WITH THE BRITISH COLUMBIA MINISTRY OF ENERGY MINES, AND PETROLEUM RESOURCES (MEMPR) UNDER A "LETTER" OF UNDERSTANDING FOR A COOPERATIVE PROJECT. THE (1985) SURVEY WAS UNDERTAKEN UNDER THE AUSPICES OF THE CANADA-BRITISH COLUMBIA MINERAL DEVELOPMENT AGREEMENT.

W.M. JOHNSON DIRECTED THE BRITISH COLUMBIA MINISTRY OF ENERGY, MINES AND PETROLEUM RESOURCES ACTIVITIES (1984,1985).

E.H.W. HORNBROOK DIRECTED THE GEOLOGICAL SURVEY OF CANADA ACTIVITIES (1984, 1985).

P.W.B. FRISKE COORDINATED THE OPERATIONAL ACTIVITIES OF THE GEOLOGICAL SURVEY OF CANADA STAFF THROUGHOUT THE SURVEY (1984,1985).

CONTRACTS LET FOR COLLECTION, SAMPLE PREPARATION AND ANALYSIS WERE THE RESPONSIBILITY OF, AND WERE SUPERVISED AND/OR MONITORED BY THE STAFF OF THE RESOURCE GEOCHEMISTRY SUBDIVISION OR THE BRITISH COLUMBIA MINISTRY AS FOLLOWS:

COLLECTION: - MCELHANNEY ENGINEERING SERVICES LTD. VANCOUVER, B.C. (1984,1985)

- H.R. SCHMITT (GSC, 1984), A. BORONOWSKI (MEMPR, 1985)

PREPARATION: - GOLDER ASSOCIATES, OTTAWA, ONTARIO (1984)

- KAMLOOPS RESEARCH AND ASSAY LABORATORY LTD., KAMLOOPS B.C.(1985)

- J.J. LYNCH (GSC, 1984), W.M. JOHNSON (MEMPR, 1985)

ANALYSIS: - BARRINGER MAGENTA LTD., REXDALE, ONTARIO (1984)

- BARRINGER MAGENTA (ALBERTA) LTD., CALGARY, ALBERTA (1984)

- BONDAR-CLEGG AND COMPANY LTD., NORTH VANCOUVER, B.C. (1985)

- CHEMEX LABS LTD., NORTH VANCOUVER, BRITISH COLUMBIA (1985)

- J.J. LYNCH (GSC, 1984), W.M. JOHNSON (MEMPR, 1985)

N.G. LUND (1984,1985), AND H.R. SCHMITT (1985) COORDINATED OPEN FILE PRODUCTION.

B. ELLIOTT (1984) AND A.C. GALLETTA (1984,1985) WERE RESPONSIBLE FOR DATA MANAGEMENT, AND FOR THE PREPARATION OF THE REGIONAL TREND MAP UTILIZING A PROGRAM DEVELOPED BY D.J. ELLWOOD.

J. YELLE (GSC) SUPERVISED MAP PREPARATION.
COMPUTING AND PLOTTING FACILITIES WERE PROVIDED BY THE COMPUTER SCIENCE CENTER, E.M.R.

OPEN FILE TEXT WAS MANUFACTURED BY K.G.CAMPBELL CORPORATION LASER PRINTING, OTTAWA.

HELICOPTER AND TRUCK SUPPORTED SAMPLE COLLECTION WAS CARRIED OUT DURING THE SUMMER OF 1984 AND 1985.

STREAM SEDIMENT AND WATER SAMPLES WERE COLLECTED AT AN AVERAGE DENSITY OF ONE SAMPLE PER 13 SQUARE KILOMETERS THROUGHOUT THE 14,800 SQUARE KILOMETERS OF THE CENTRAL B.C SURVEY AREA (NTS 93G).

SAMPLE SITE REPLICATE SAMPLES WERE ROUTINELY COLLECTED IN EACH BLOCK OF TWENTY SAMPLE NUMBERS.

FIELD DRIED SAMPLES WERE AIR-DRIED, SIEVED THROUGH AN 80 MESH SCREEN (1984,1985), AND BALL MILLED (1984 SAMPLES ONLY). AT THIS TIME, CONTROL REFERENCE AND BLIND DUPLICATE SAMPLES WERE INSERTED INTO EACH BLOCK OF TWENTY SEDIMENT SAMPLES. FOR THE WATER SAMPLES, ONLY CONTROL REFERENCE SAMPLES WERE INSERTED INTO THE BLOCK. THERE WERE NO BLIND DUPLICATE WATER SAMPLES.

ON RECEIPT, FIELD AND ANALYTICAL DATA WERE PROCESSED WITH THE AID OF COMPUTERS.

THE FIELD DATA WERE RECORDED BY THE FIELD CONTRACT STAFF ON STANDARD STREAM WATER AND SEDIMENT FIELD CARDS (REV. 74) USED BY THE GEOLOGICAL SURVEY OF CANADA (GARRETT, 1974).

THE SAMPLE SITE POSITIONS WERE MARKED IN THE FIELD ON APPROPRIATE 1/250,000 SCALE NTS MAPS

THESE MAPS WERE DIGITIZED AT THE GEOLOGICAL SURVEY IN OTTAWA TO OBTAIN THE SAMPLE SITE UTM COORDINATES.

THE SAMPLE SITE COORDINATES WERE CHECKED AS FOLLOWS: A SAMPLE LOCATION MAP WAS PRODUCED ON A CALCOMP 1051 DRUM PLOTTER USING THE DIGITIZED COORDINATES; THE FIELD CONTRACTOR'S SAMPLE LOCATION MAP WAS THEN OVERLAYED WITH THE CALCOMP MAP; THE TWO SETS OF POINTS WERE CHECKED FOR COINCIDENCE. THE DOMINANT ROCK TYPES IN THE STREAM CATCHMENT BASINS WERE IDENTIFIED ON APPROPRIATE GEOLOGICAL MAPS USED AS THE BEDROCK GEOLOGICAL BASE ON RGR MAPS.

THOROUGH INSPECTIONS OF THE FIELD AND ANALYTICAL DATA WERE MADE TO CHECK FOR ANY MISSING INFORMATION AND/OR GROSS ERRORS.

QUALITY CONTROL AND MONITORING OF THE GEOCHEMICAL DATA WAS UNDERTAKEN BY A STANDARD METHOD USED BY THE RESOURCE GEOCHEMISTRY SUBDIVISION AT THE GEOLOGICAL SURVEY OF CANADA.

FOR THE DETERMINATION OF ZN, CU, PB, NI, CO, AG, MN, FE, CD, AND AS A 1 GRAM SAMPLE WAS REACTED WITH 3 ML CONC. HNO3 IN A TEST TUBE OVERNIGHT AT ROOM TEMPERATURE.

AFTER DIGESTION, THE TEST TUBE WAS IMMERSED IN A HOT WATER BATH AT ROOM TEMPERATURE AND BROUGHT UP TO 90C AND HELD AT THIS TEMPERATURE FOR 30 MINUTES WITH PERIODIC SHAKING. 1 ML CONC. HCL WAS ADDED AND HEATING WAS CONTINUED FOR ANOTHER 90 MINUTES.

THE SAMPLE SOLUTION WAS THEN DILUTED TO 20 ML WITH METAL FREE WATER AND MIXED. ZN, CU, PB, NI, CO, AG, MN, FE AND CD WERE DETERMINED BY ATOMIC ABSORPTION SPECTROSCOPY USING AN AIR-ACETYLENE FLAME.

BACKGROUND CORRECTIONS WERE MADE FOR PB, NI, CO, AG AND CD.
AS WAS DETERMINED BY ATOMIC ABSORPTION USING A HYDRIDE EVOLUTION METHOD
WHEREIN THE HYDRIDE (ASH3) IS EVOLVED, PASSED THROUGH A HEATED QUARTZ
TUBE IN THE LIGHT PATH OF AN ATOMIC ABSORPTION SPECTROPHOTOMETER.
THE METHOD IS DESCRIBED BY ASLIN (1976).

MOLYBDENUM AND VANADIUM (1984) WERE DETERMINED BY ATOMIC ABSORPTION SPECTROSCOPY USING A NITROUS OXIDE ACETYLENE FLAME.

A 0.5 GRAM SAMPLE WAS REACTED WITH 1.5 ML CONCENTRATED HN03 AT 90C FOR 30 MINUTES.

AT THIS POINT 0.5 ML CONCENTRATED HCL WAS ADDED AND THE DIGESTION WAS CONTINUED AT 90C FOR AN ADDITIONAL 90 MINUTES.

AFTER COOLING, $8\,$ ML of 1250 PPM AL SOLUTION WERE ADDED AND THE SAMPLE SOLUTION WAS DILUTED TO $10\,$ ML BEFORE ASPIRATION.

MERCURY WAS DETERMINED BY THE HATCH AND OTT PROCEDURE WITH SOME MODIFICATIONS. THE METHOD IS DESCRIBED BY JONASSON ET AL. (1973).

A 0.5 GRAM SAMPLE WAS REACTED WITH 20 ML CONCENTRATED HNO3 AND 1 ML CONCENTRATED HCL IN A TEST-TUBE FOR 10 MINUTES AT ROOM TEMPERATURE PRIOR TO 2 HOURS OF DIGESTION WITH MIXING AT 90C IN A HOT WATER BATH.

AFTER DIGESTION, THE SAMPLE SOLUTIONS WERE COOLED AND DILUTED TO 100 ML WITH METAL FREE WATER.

THE HG PRESENT WAS REDUCED TO THE ELEMENTAL STATE BY THE ADDITION OF 10 ML W/V SNSO4 IN M H2SO4.

THE HG VAPOUR WAS THEN FLUSHED BY A STREAM OF AIR INTO AN ABSORPTION CELL MOUNTED IN THE LIGHT PATH OF AN ATOMIC ABSORPTION SPECTROPHOTOMETER. ABSORPTION MEASUREMENTS WERE MADE AT 253.7 NM.

LOSS ON IGNITION (1984) WAS DETERMINED USING A 500 MG SAMPLE. THE SAMPLE, WEIGHED INTO 30 ML BEAKER, WAS PLACED IN A COLD MUFFLE FURNACE AND BROUGHT UP TO 500C OVER A PERIOD OF 2-3 HOURS. THE SAMPLE WAS LEFT AT THIS TEMPERATURE FOR 4 HOURS, THEN ALLOWED TO COOL TO ROOM TEMPERATURE FOR WEIGHING.

URANIUM (1985) WAS DETERMINED USING A NEUTRON ACTIVATION METHOD WITH DELAYED NEUTRON COUNTING. WITH THE EXCEPTION OF IRRADIATION FACILTY, THE METHOD IS VERY SIMILAR TO THAT USED BY AECL IN PREVIOUS YEARS. A DETAILED DESCRIPTION OF WHICH IS PROVIDEDBY BOULANGER ET AL (1975).

A TWO GRAM SAMPLE WAS IRRADIATED FOR 10 SECONDS. THE OPERATING FLUX WAS $10 \times 10 \times 13$ NEUTRONS/SQUARE CM/SECOND. AFTER A 10 SECOND DELAY, THE SAMPLE WAS COUNTED FOR 10 SECONDS. THE COUNTING EQUIPMENT WAS OF AECL DESIGN. CALIBRATION WAS DONE TWICE A DAY OR AS REQUIRED. ONE STANDARD WAS ANALYSED AFTER EVERY 20 SAMPLES.

URANIUM (1984) WAS DETERMINED USING A NEUTRON ACTIVATION METHOD WITH DELAYED NEUTON COUNTING. A DETAILED DESCRIPTION OF THE METHOD IS PROVIDED BY BOULANGER ET AL(1975). IN BRIEF, A 1 GRAM SAMPLE IS WEIGHED INTO A 7 DRAM POLYETHYLENE VIAL, CAPPED AND SEALED.

THE IRRADIATION IS PROVIDED BY THE SLOWPOKE REACTOR WITH AN OPERATING FLUX OF 10** 12 NEUTRONS/SO.CM./SEC.

THE SAMPLES ARE PNEUMATICALLY TRANSFERRED FROM AN AUTOMATIC LOADER TO THE REACTOR, WHERE EACH SAMPLE IS IRRADIATED FOR 60 SECONDS.

AFTER IRRADIATION, THE SAMPLE IS AGAIN TRANSFERRED PNEUMATICALLY TO THE COUNTING FACILITY WHERE AFTER A 10 SECOND DELAY THE SAMPLE IS COUNTED FOR 60 SECONDS WITH SIX BF3 DETECTOR TUBES EMBEDDED IN PARAFFIN. FOLLOWING COUNTING, THE SAMPLES ARE AUTOMATICALLY EJECTED INTO A SHIELDED STORAGE CONTAINER.

CALIBRATION IS CARRIED OUT TWICE A DAY AS A MINUMUM, USING NATURAL MATERIALS OF KNOWN URANIUM CONCENTRATION.

TUNGSTEN WAS DETERMINED AS FOLLOWS: A 0.2 GRAM SAMPLE OF STREAM SEDIMENT WAS FUSED WITH 1 GRAM K2S2O7 IN A RIMLESS TEST TUBE AT 575C FOR 15 MINUTES IN A FURNACE. THE COOLED MELT WAS THEN LEACHED WITH 10 ML CONCENTRATED HCL IN A WATER BATH HEATED TO 85C. AFTER THE SOLUBLE MATERIAL HAD COMPLETELY DISSOLVED, THE INSOLUBLE MATERIAL WAS ALLOWED TO SETTLE AND AN ALIQUOT OF 5 ML WAS TRANSFERRED TO ANOTHER TEST TUBE. 5 ML OF 20% SNCL2 SOLUTION WERE THEN ADDED TO THE SAMPLE ALIQUOT, MIXED AND HEATED FOR 10 MINUTES AT 85C IN A HOT WATER BATH. A 1 ML ALIQUOT OF DITHIOL SOLUTION (1% DITHIOL IN ISO-AMYL ACETATE) WAS ADDED TO THE TEST SOLUTION AND THE TEST SOLUTION WAS THEN HEATED FOR 4-6 HOURS AT 80-85C IN A HOT WATER BATH. THE TEST SOLUTION WAS THEN REMOVED FROM THE HOT WATER BATH, COOLED AND 2.5 ML OF KEROSENE ADDED TO DISSOLVE THE GLOBULE, THE COLOUR INTENSITY OF THE KEROSENE SOLUTION WAS MEASURED AT 630 NM USING A SPECTROPHOTOMETER.

A DETAILED DESCRIPTION OF THE METHOD IS GIVEN BY QUIN AND BROOKS(1972)

BARIUM WAS DETERMINED AS FOLLOWS: A 0.25 GRAM SAMPLE WAS HEATED WITH 5 ML CONC. HF, 5 ML CONC. HCLO4 AND 2 ML CONC. HNO3 TO FUMES OF HCLO4; 3 ML OF CONC. HCLO4 WERE ADDED AND HEATED TO LIGHT FUMES; 5 ML OF WATER WERE ADDED AND THE SOLUTION WAS TRANSFERRED TO A CALIBRATED TEST TUBE AND DILUTED TO 25 ML WITH WATER. BARIUM WAS DETERMINED BY ATOMIC ABSORPTION SPECTROSCOPY USING A NITROUS OXIDE ACETYLENE FLAME.

ANTIMONY (1984-1985) WAS DETERMINED IN STREAM SEDIMENTS AS DESCRIBED BY ASLIN(1976). A 500 MG SAMPLE IS PLACED IN A TEST TUBE; 3 ML CONCENTRATED HNO3 AND 9 ML CONCENTRATED HCL ARE ADDED AND THE MIXTURE IS ALLOWED TO STAND OVERNIGHT AT ROOM TEMPERATURE. THE MIXTURE IS HEATED SLOWLY TO 90C AND MAINTAINED AT THIS TEMPERATURE FOR AT LEAST 90 MINUTES. THE SOLUTION IS COOLED AND DILUTED TO 10 ML. A 400 UL ALIQUOT OF THIS TEST SOLUTION IS REMOVED AND DILUTED TO 10 ML WITH 1.8 M HCL. THE ANTIMONY IN AN ALIQUOT OF THIS DILUTE SOLUTION IS THEN DETERMINED BY HYDRIDE EVOLUTION-ATOMIC ABSORPTION SPECTROMETRY.

FLUORIDE IN STREAM WATER SAMPLES WAS DETERMINED USING A FLUORIDE ELECTRODE. PRIOR TO MEASUREMENT AN ALIQUOT OF THE SAMPLE WAS MIXED WITH AN EQUAL VOLUME OF TISAB II SOLUTION (TOTAL IONIC STRENGTH ADJUSTMENT BUFFER). THE TISAB II BUFFER SOLUTION WAS PREPARED AS FOLLOWS: TO 50 ML METAL FREE WATER ADD 57 ML GLACIAL ACETIC ACID, 58 GM NACL AND 4 GM CDTA (CYCLOHEXYLENE DINITRILO TETRAACETIC ACID). STIR TO DISSOLVE AND COOL TO ROOM TEMPERATURE. USING A PH METER, ADJUST THE PH BETWEEN 5.0 AND 5.5 BY SLOWLY ADDING 5 M NAOH SOLUTION. COOL AND DILUTE TO ONE LITER IN A VOLUMETRIC FLASK.

HYDROGEN ION ACTIVITY (PH) WAS MEASURED WITH A COMBINATION GLASS-CALOMEL ELECTRODE AND A PH METER.

URANIUM IN WATERS WAS DETERMINED BY A LASER-INDUCED FLUOROMETRIC METHOD USING A SCINTREX UA-3 URANIUM ANALYSER.

A COMPLEXING AGENT, KNOWN COMMERCIALLY AS FLURAN AND COMPOSED OF SODIUM PYROPHOSPHATE AND SODIUM MONOPHOSPHATE, (HALL, G.E.M., 1979) IS ADDED TO PRODUCE THE URANYL PYROPHOSATE SPECIES WHICH FLUORESCES WHEN EXPOSED TO THE LASER.

SINCE ORGANIC MATTER IN THE SAMPLE CAN CAUSE UNPREDICABLE BEHAVIOUR, A STANDARD ADDITION METHOD WAS USED.

FURTHER, THERE HAVE BEEN INSTANCES AT THE G.S.C. WHERE THE REACTION OF URANIUM WITH FLURAN IS EITHER DELAYED OR SLUGGISH; FOR THIS REASON AN ARBITRARY 24 HOUR TIME DELAY BETWEEN THE ADDITION OF THE FLURAN AND THE ACTUAL READING WAS INCORPORATED INTO THIS METHOD.

IN PRACTICE,500 UL OF FLURAN SOLUTION WERE ADDED TO A 5 ML SAMPLE AND ALLOWED TO STAND FOR 24 HOURS. AT THE END OF THIS PERIOD FLUORESCENCE READINGS WERE MADE WITH THE ADDITION OF 0.0, 0.2 AND 0.4 PPB U.

FOR HIGH SAMPLES THE ADDITIONS WERE 0.0, 2.0 AND 4.0 (20 UL ALIQUOTS OF EITHER 55 OR 550 PPB U WERE USED).

ALL READINGS WERE TAKEN AGAINST A SAMPLE BLANK.

THE FOLLOWING TABLES DISPLAY THE DATA RECORD FORMAT SPECIFICATIONS. THE DETECTION LIMITS OF THE ANALYTICAL METHODS ARE GIVEN. THE SECOND FIGURE UNDER THE DETECTION LIMIT HEADING IS USED ARBITRARILY TO DENOTE VALUES BELOW THE DETECTION LIMIT(USUALLY 1/2 DETECTION LIMIT)

ELEMENT	CARD	COLUMNS
FIELD		
MAP	1	01-06
ID	1	07-12
UTM ZONE	1	13-14
UTM EAST (METER)	1	15-20
UTM NORTH (METER)	1	21-27
ROCK TYPE	1	28-31
SAMPLE MATERIAL	1	32
STREAM WIDTH (DECIMETER	.) 1	33-35
STREAM DEPTH (DECIMETER	.) 1	36-38
REPLICATE STATUS	1	39-40
CONTAMINATION	1	41
BANK TYPE	1	42
WATER COLOUR	1	43
FLOW RATE	1	44
SEDIMENT COLOUR	1	45
SAMPLE COMPOSITION	1	46-48
PRECIPITATE IN STREAM	1	49
DISTINCTIVE PRECIPITATE	1	50
GENERAL PHYSIOGRAPHY	1	55
DRAINAGE PATTERN	1	56
STREAM TYPE	1	57
STREAM CLASS	1	58
SOURCE OF WATER	1	59
AGE	1	70-71

THE ANALYTICAL DATA WERE RECORDED AS FOLLOWS:

ELEME		UNITS	CARD	COLUMNS	DETECTION	LIMIT
SEDIMENT						
	ZN	PPM	2	21-25	2	1
	CU	PPM	2	26-30	2	1
	PB	PPM	2	31-35	2	1
]	NI	PPM	2	36-40	2	1
	C0	PPM	2	41-45	2	1
	AG	PPM	2	46-50	0.2	0.1
j	MN	PPM	2	51-55	5	2
	AS	PPM	2	56-60	1.0	0.5
I	MO	PPM	2	61-65	2	1
	FE	PCT	2	66-70	0.02	0.01
	HG	PPB	2	71-75	10	5
	LOI	PCT	2	76-80	1.0	0.5
1	U	PPM	3	21-25	0.5	0.2
,	V	PPM	3	31-35	5	2
	CD	PPM	3	36-40	0.2	0.1
1	W	PPM	3	46-50	2	1
	SB	PPM	3	56-60	0.2	0.1
	BA	PPM	3	61-65	40	20
WATER						
	F	PPB	4	26-30	20	10
	PH		4	31-35		
i	U	PPB	4	36-40	0.05	0.02

REFERENCES

- ASLIN, G.E.M. (1976) THE DETERMINATION OF ARSENIC AND ANTIMONY IN GEOLOGICAL MATERIALS BY FLAMELESS ATOMIC ABSORPTION SPECTROPHOTOMETER JOURNAL OF GEOCHEMICAL EXPLORATION, VOL. 6, Pp. 321-330.
- BOULANGER, A., EVANS, D.J.R. AND RABY, B.F. (1975) URANIUM ANALYSIS BY NEUTRON ACTIVATION DELAYED NEUTRON COUNTING: PROC. OF THE 7TH ANNUAL SYMP. OF CANADIAN MINERAL ANALYSTS. THUNDER BAY, ONTARIO, SEPT. 22-23, 1975.
- FICKLIN, W.H. (1970) A RAPID METHOD FOR THE DETERMINATION OF FLUORIDE IN ROCKS AND SOILS, USING AN ION SELECTIVE ELECTRODE. U.S. GEOL. SURV. PAPER 700C PP. C186-188.
- GARRETT, R.G. (1974) FIELD DATA AQUISITION METHODS FOR APPLIED GEOCHEMICAL SURVEYS AT THE GEOLOGICAL SURVEY OF CANADA: GEOL SURV. CAN. PAPER 74-52.
- HALL, G.E.M. (1979) A STUDY OF THE STABILITY OF URANIUM IN WATERS COLLECTED FROM VARIOUS GEOLOGICAL ENVIRONMENTS IN CANADA; IN CURRENT RESEARCH, PART A, GEOL. SURV. CAN. PAPER 79-1A, P. 361-365.
- JONASSON, I.R., LYNCH, J.J. AND TRIP, L.J. (1973) FIELD AND LABORATORY METHODS USED BY THE GEOLOGICAL SURVEY OF CANADA IN GEOCHEMICAL SURVEYS: NO. 12, MERCURY IN ORES, ROCKS, SOILS, SEDIMENTS AND WATER: GEOL. SURV. CAN. PAPER 73-21.
- QUIN, B.F. AND BROOKS, R.R. (1972) THE RAPID DETERMINATION OF TUNGSTEN IN SOILS, STREAM SEDIMENTS, ROCKS AND VEGETATION. ANAL.CHIM. ACTA. 58 PP 301-309.

DATA LIST LEGEND

MAP-	NATIONAL TOPOGRAPHIC SYSTEM(NTS) - LETTERED QUADRANGLE (SCALE 1:50000). PART OF SAMPLE NUMBER REMAINDER OF SAMPLE NUMBER- YEAR(2), FIELD CREW(1), SAMPLE SEQUENCE NUMBER(3)
ZN- FAST-	UNIVERSAL TRANSVERSE MERCATOR (UTM) COORDINATE SYSTEM- SAMPLE COORDINATES ZONE EASTING (METERS) NORTHING (METERS)
ROCK TYPE-	MAJOR ROCK TYPE OF STREAM CATCHMENT AREA STRATIGRAPHIC AGE OF ROCK TYPE
WD- DT-	WIDTH OF STREAM(DECIMETER) AT NEAREST SAMPLE SITE DEPTH OF STREAM SAMPLED TO NEAREST DECIMETER
SAMP-	TYPE OF MATERIAL SAMPLED
RP ST-	REPLICATE STATUS- RELATIONSHIP OF SAMPLE WITH RESPECT TO OTHERS WITHIN THE SURVEY
CONT-	CONTAMINATION
WCOL-	WATER COLOUR AND SUSPENDED LOAD
RATE-	WATER FLOW RATE
SCOL-	PREDOMINANT SEDIMENT COLOUR
SMP CMP-	SAMPLE COMPOSITION- BULK MECHANICAL COMPOSITION OF SAND, FINES AND ORGANICS RESPECTIVELY
PPPS-	PRECIPITATE OR STAIN ON SEDIMENTS AT SAMPLE SITE
PRPB-	DISTINCTIVE PRECIPITATE, STAIN, WEATHERING, BLOOMS ON ROCKS IN IMMEDIATE CATCHMENT AREA
PHYS-	GENERAL PHYSIOGRAPHY
PATT-	DRAINAGE PATTERN
TYPE-	STREAM TYPE
CLSE-	STREAM CLASS
SRCE-	SOURCE OF WATER

REGIONAL STREAM SEDIMENT AND WATER GEOCHEMICAL DATA, BRITISH COLUMBIA 1985, GSC-OF 1214, NGR 79-1985, NTS 93G

ROCK TYPE/AGE:

CENOZOIC:

- (TILL 44) TILL GRAVEL, SAND, SILT, ALLUVIUM
- (BSLT 42) OLIVINE BASALT FLOWS, BRECCIA AND TUFF
- (SNDS 42) SANDSTONE, SHALE, CONGLOMERATE, DIATOMITE, LIGNITE
- (ANDS 42) ENDAKO GROUP: ANDESITE, BASALT, DACITE
- (CGLM 42) CONGLOMERATE, SANDSTONE, SHALE, TUFF, BRECCIA

MESOZOIC - CENOZOIC:

- (RYLT 41) OOTSA LAKE GROUP:RHYOLITEDACITE,TRACHYTE, SANDSTONE,SHALE,CONGLOMERATE
- (ANDS 36) ANDESITE, TUFF, BRECCIA, ARGILLITE, ARKOSE, CONGLOMERATE
- (CGLM 36) SKEENA GROUP: CONGLOMERATE, GREYWACKE, SHALE, COAL, VOLCANIC BRECCIA
- (ANDS 34) HAZELTON GROUP (PART) UNDIVIDED: BASALT,
 ANDESITE, TUFF, BRECCIA, GREYWACKE, MUDSTONE,
 CONGLOMERATE
- (SHLE 34) SHALE, GREYWACKE, CONGLOMERATE
- (ANDS 33) TAKLA GROUP: ANDESITE, BASALT, TUFF, BRECCIA, CONGLOMERATE, GREYWACKE, SHALE, LIMESTONE
- (LMSN 32) LIMESTONE
- (PLLT 32) BLACK PHYLLITE, SILTSTONE, LIMESTONE, QUARTZITE

PALEOZOIC:

- (CHRT 23) CACHE CREEK GROUP:RIBBON CHERT, BLACK ARGILLITE, LIMESTONE, GREENSTONE
- (BSLT 21) SLIDE MOUNTAIN GROUP: BASALT, BRECCIA, TUFF, CHERT, ARGILLITE, SANDSTONE, LIMESTONE, CONGLOMERATE

PROTEROZOIC:

- (SNDS 04) KAZA GROUP: SANDSTONE, CONGLOMERATE, GRIT, PHYLLITE, SCHIST, AMPHIBOLITE, MARBLE, GNEISS
- (QTMZ 36) NAVER INTRUSIONS:QUARTZ MONZONITE, SYENITE, MONZONITE, GRANODIORITE, DIORITE
- (GRDR 32) TAKOMKANE BATHOLITH AND BODIES OF SIMILAR
 AGE AND LITHOLOGY: GRANODIORITE, QUARTZ
 DIORITE, QUARTZ MONZONITE
- (SRPN 31) TREMBLEUR INTRUSIONS AND SIMILAR BODIES:
 PERIDOTITE, DUNITE, PYROXENITE, SERPENTINE

AGE:	RATE:	TYPE:
04 - HADRYNIAN	0 - ZERO	0 - UNDEFINED
21 - MISSISSIPPIAN	1 - SLOW	1 - PERMANENT, CONTINUOUS
23 - PENNSYLVANIAN AND PERMIAN	2 - MODERATE	2 - INTERMITTENT
31 - PERMIAN AND/OR TRIASSIC	3 - FAST	3 - RE-EMERGENT, DISCONTINUOUS
32 - TRIASSIC		,
	SCOL:	CLSE:
34 - JURASSIC	0 - UNKNOWN	2 - SECONDARY
36 - CRETACEOUS	1 - RED, BROWN	3 - TERTIARY
41 - UPPER-CRETACEOUS-LOWER TERTIARY (NTS 93G)		4 - QUARTERNARY
42 - TERTIARY	3 - BLACK	
44 - PLEISTOCENE AND RECENT	5 - GREEN	SRCE:
	6 - GREY	0 - UNKNOWN
SAMP:	8 - BUFF TO BROWN	1 - GROUNDWATER
1 - STREAM BED SEDIMENT		3 - RECENT PRECIPITAION
6 - SIMULTANEOUS STREAM WATER		
AND SEDIMENT	SAM CMP: - PORTION OF EACH C	COMPONENT IS INDICATED
	AS A FRACTION OF	THE TOTAL OF ALL THREE
RP ST:	COLUMNS	
00 - ROUTINE REGIONAL SAMPLE	EXAMPLE:	
10 - FIRST OF FIELD DUPLICATE	013 NO SAND, 25% F	
20 - SECOND OF FIELD DUPLICATE		FINES, 40% ORGANICS
	030 NO SAND, 100%	FINES, NO ORGANICS
CONT:	111 33% SAND, 33%	FINES, 33% ORGANICS
0 - NONE		
1 - POSSIBLE	PPPS:	
2 - PROBABLE	0 - NONE	
3 - DEFINITE	1 - RED, BROWN	
6 - AGRICULTURAL	2 - WHITE, BUFF	
8 - FORESTRY ACTIVITY	3 - BLACK	
9 -BURNED AREAS	2222	
	PRPB:	
BANK:	0 - FEATURELESS	
0 - UNDEFINED UNCONSOLIDATED MATERIAL	2 - WHITE, BUFF	
1 - ALLUVIAL 2 - COLLUVIAL	PHYS:	
3 - GLACIAL TILL, TILLITE 4 - GLACIAL OUTWASH, MORAINE	1 - MUSKEG, SWAMPLAND 2 - PENEPLAIN, PLATEAU	
5 - BARE ROCK	3 - HILLY, UNDULATING	
6 - TALUS, SCREE	4 - MOUNTAINOUS, MATURE	
7 - ORGANIC PREDOMINANT	4 MOONTAINOOS, MATORE	
/ OROMIC TREBOTTIVING	PATT:	
WCOL:	0 - POORLY DEFINED, HAPHAZ	ARD
0 - CLEAR	1 - DENDRITIC	
1 - BROWN TRANSPARENT	2 - HERRINGBONE	
2 - WHITE CLOUDY	5 - DISCONTINUOUS SHIELD	TYPE
3 - BROWN CLOUDY	(CHAINS OF LAKES AND	
	6 - BASINAL	

- ZN- ZINC BY ATOMIC ABSORPTION SPECTROSCOPY (PPM)
- CU- COPPER BY ATOMIC ABSORPTION SPECTROSCOPY (PPM)
- PB- LEAD BY ATOMIC ABSORPTION SPECTROSCOPY (PPM)
- NI- NICKEL BY ATOMIC ABSORPTION SPECTROSCOPY (PPM)
- CO- COBALT BY ATOMIC ABSORPTION SPECTROSCOPY (PPM)
- AG- SILVER BY ATOMIC ABSORPTION SPECTROSCOPY (PPM)
- MN- MANGANESE BY ATOMIC ABSORPTION SPECTROSCOPY (PPM)
- AS- ARSENIC BY COLOURIMETRY (PPM)
- MO- MOLYBDENUM BY ATOMIC ABSORPTION SPECTROSCOPY (PPM)
- FE- IRON BY ATOMIC ABSORPTION SPECTROSCOPY(%)
- HG- MERCURY BY FLAMELESS SPECTROSCOPY (PPB)
- LOI- LOSS ON IGNITION BY WEIGHT DIFFERENCE (%)
- U- URANIUM BY DELAYED NEUTRON ACTIVATION (PPM)
- V- VANADIUM BY ATOMIC ABSORPTION SPECTROSCOPY (PPM)
- CD- CADMIUM BY ATOMIC ABSORPTION SPECTROSCOPY (PPM)
- W- TUNGSTEN BY COLORIMETRY USING DITHIOL (PPM)
- SB- HYDRIDE EVOLUTION-ATOMIC ABSORPTION
 - SPECTROMETRY (PPM)
- BA- BARIUM BY ATOMIC ABSORPTION (PPM)
- F-W- FLUORINE IN WATER BY FISSION TRACK (PPB)
- PH- PH BY COMBINATION GLASS-CALOMEL ELECTRODE
- U-W- URANIUM IN WATERS BY SCINTREX (PPB)